MINUTES OF THE 21st MEETING OF THE BOARD OF STUDIES OF THE CHEMISTRY DEPARTMENT HELD ON 9th December 2023

The meeting of the Board of Studies of the Department of Chemistry was held on 9th December2023 at 10 am in the Department Library. In the beginning, the Chairperson introduced and welcomed the members. The following agendas were taken up for discussion.

Minute No	Minutes
1.	Confirmation of the Minutes of the 20th BoS meeting held on 24th June 2023
	The Chairperson Dr.G.Allen Gnana Raj read the minutes of the BOS meeting and it was confirmed.
2.	Action Taken Report on the Minutes of the 20th BOS meeting
	The Chairperson Dr. G. Allen Gnana Raj presented the following action taken during the previous BOS
	To increase the percentage of course outcome, selected modules are modified.
	Questions are ensured that they are in the understanding level by the students.
	The members of BOS carefully perused the presented action by the Chairperson and expressed their satisfaction and appreciation.
	Review of Academic and Administrative Audit (AAA) Report
3.	 The Chairperson Dr. G.Allen Gnana Raj presented the AAA report criteria-wise and the members of BOS carefully reviewed the report and appreciated the following. Awards given to the high impact publication for staff and research scholars. Awards given to the best paper presentation for the PG students and research scholars.

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Review of Feedback Report on curriculum obtained from 1) Students

2) Teachers 3) Employers 4) Alumni

The Chairperson presented the Feedback Report obtained from 1) Students 2) Teachers 3) Employers 4) Alumni on the curriculum of B.Sc. and M.Sc Chemistry from the

academic year 2022-2023.

The members of BOS carefully reviewed the Feedback Report received from the stake holders and proposed the following actions to be taken in the curriculum.

- Remedial classes should be conducted regularly for the slow learners.
- Advanced learners should be identified and motivated for the competitive examinations.

5. Introduction of new courses from the academic year 2023-24

As per the suggestion of Dr.J. Helen Retna Monica and Dr. T.S. Abbs Fenn Regie the following changes were done in both UG and PG programme.

Dr. G. Allen Gana Raj proposed the introduction of the following new courses in the curriculum of B.Sc./M.Sc. Chemistry from the academic year 2023-'24.

1. Name of the Course with local/regional/national/global relevance

Name of the Course	Code	Relevance to local, national, regional and global development needs
1.RubberTechnology-Skill	en Gan	Regional
Enhancement Course		National
2.Entepereneurial Skills in Chemistry	lue.	TVALIGIAL
3. Material Science	•	National
4. Bio inorganic Chemistry	-	Global
5. Chemistry of Natural Products	-	Global

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Suggestions and recommendations for changes in UG/PG Chemistry syllabus 2023-24

As per the recommendation of Dr. T.S. Abbs Fenn Regie we have made few changes in the UG curriculum frame work. He suggested to introduce General Chemistry II in the second semester and General Chemistry III in the third semester.

As per the suggestion of Dr.J. Helen Retna Monica we have introduced Material Science and Bioinorganic chemistry as elective papers in the PG Programme. Advanced topics in Organic , Physical and Inorganic Chemistry were introduced in the Core papers. Since experimental learning is more effective she suggested to give equal importance to both theory and practical's.

6. Revision of syllabus of existing courses from the academic year 2023-2024

CHANGES MADE IN THE UG PROGRAMME

No	Seme ster	Name of the paper	Observations/Suggestions /Recommendations	Relevance toGlobal/ National/ Regional	%of Change
1	I	General Chemistry II	Unit I- Acids, bases & Ionic equilibria Unit II- Chemistry of s –Block Elements Unit III- Chemistry of p-Block Elements Unit IV- Hydrocarbon Chemistry-I Unit V- Colloids & Surface Chemistry	Global	100% change
2	п	Organic Chemistry I	I-Streochemistry II-Chemistry of Nitrogen Compounds -I III- Chemistry of Nitrogen Compunds-II IV-Heterocylic compunds V-Six-membered heterocyclic compounds	Global	20% change
3	III	General Chemistry III	Unit I- Gaseous state Unit II- Liquid State & Solid States Unit III- Nuclear Chemistry Unit IV- Halogen derivatives	Global	100% change

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			Unit V- Phenols and Aromatic alcohols	T	
4		Physical	Unit I- Thermodynamics I	Global	shifted
	III	Chemistry I	Unit II- Thermodynamics II		from
			Unit III- Thermodynamics III		Sem V
			Unit IV- Phase Rule -I		to III
		W 5000	Unit V- Phase Rule- II	Company let	
5	IV	Inorganic	I- Hydrogen and Oxygen compounds	Global	25%
		Chemistry- I	II- Halogen compounds		
			III- Carbon group element		
			IV- d-Block element		
			V- f-Block element		
	V	Organic	Unit I- Alkalodis and Terpenes	Global	40%
6		Chemistry -II	Unit II-Carbohydarates		change
	TOTAL S		Unit III-Molecularrear rearrangements		
		1	Unit IV -Ethers, Thioethers & Epoxides		
			UnitV – Carboxylie Acids	101	
7	VI	Inorganic	Unit I- Coordination Compounds-I	Global	20%
		Chemistry II	Unit I- Coordination Compounds-II		Change
			Unit I- Coordination Compounds-III		
			Unit I- Coordination Compounds-IV		S. Whange
			Unit I- Inorganic Chemistry of Biological		
			System		
8	VI	Physical	Unit I- Chemical Kinetics – delete 1.12	Global	10 %
		Chemistry -II	Unit II- Molecular Symmetry and Group	180	change
			Theory		
			Unit III- Electro Chemistry -I	1000	
			Unit IV- Electro Chemistry -II		
			Unit V- Electro Chemistry -III	1991	fr-Sage
			Elective	7	
9	V	Polymer	II.Polymer and its types II.Tecniques of	Regional	30%
		Chemistry	Polymerization,		

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			Synthesis and Characterisation III.Synthetic and Natural Polymers IV.Physical State and Biomedical Application of Polymers V. Properties & Processing of Polymers		
10	V	Industrial Chemistry	I – Fuels II – Cosmetics III - Sugar industry, food preservation and processing IV - Abrasives, Leather and Paper industry V - Lubricants and Cement	Regional	40%

Practicals

1.	11	Qualiitative Organic Analysis	Global	No Change
		Dr. Dried model Control (Sp.		
2	111	Qualitative analysis of simple inorganic salt	Global	50% change
3	1V	Qualitative analysis of Salt mixture	Global	No change
4	V	Organic Estimation and determination of Physical constant	Global	20%change
3	V	Gravimeteric Estimation & Inorganic Preperation	Global	No change
4	V	Project	Regional	
5	VI	Physical Chemistry Experiments	Global	No change

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Multidisiciplinary Courses

1.	ı	FOOD CHEMISTRY	Regional	50%
		I. Food Adulteration		
		II. Food Poison		
	1 1	III. Food Additives		
		IV.Beverages		
		V.Edible Oils		
3.	III	DAIRY CHEMISTRY I. Properties of milk	Regional	60%
		II. Processing of milk		
		III. Milk Products I		
	100	IV. Milk Products II	No. of	12 x
		V. Milk Lipids / Enzymes		
4.	IV	CHEMISTRY IN EVERYDAY LIFE 1.Pollution	National	40%
		II. Building Materials and plastics		
		III. Food and Cosmetics		
		IV. Explosives and Fuels		
	1	V.Pharmaceutical drugs		
5.	V	RUBBER TECHNOLOGY I.Introduction of natural rubber	Regional	100%
		II. Methods of rubber processing III.Rubber Tapping and Rubber Latex IV. Rubber additives		
		V. Applications of Rubber products		

Skill Enhancement Course

1	III	COSMETICS AND	Unit-I Skin Care		100%
		PERSONAL	Unit-II Hair Care and Dental Care		change
		GROOMING	Unit-III Make Up	Global	

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: = *			Unit -IV Perfumes Unit -V Beauty treatments		
2	IV	ENTREPRENEURI AL SKILLS IN CHEMISTRY	Unit-1 Food chemistry and Dyes Unit-II Hands on Experience	Regional	100% change
3	v	PESTICIDES IN CHEMISTRY	Unit-IToxicity of pesticides Insecticides Unit-II Pesticides residues Unit-III Bio pesticides	Regional	100% change
4	VI	INSTRUMENTAL METHODS OF CHEMICAL ANALYSIS	Unit-1 Quantitative and Qualitative aspects of Analysis Unit-2 Atomic Absorption Spectroscopy Unit-3 UV-Visible and IR Spectroscopy Unit -4 Thermal and Electro- Analytical Methods of Analysis Unit -5 Separation and Purification techniques	National	100% change

Minor Stream (Allied) Course

			I – Water and Ozone	Regional	40%
1 II	п	Allied II – Carbohydrat	II – Carbohydrates		
		Chemistry	III – Amino acids and Essential elements		
			IV – Polymer Chemistry	1 .5 F	
		n Real proof of	V - Photochemistry	regard Marie	
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CHANGES MADE IN THE PG PROGRAMME

Semester	Name of the paper	Observations/Suggestions /Recommendations	Relevance toGlobal/ National/ Regional	% of Change
I	Organic chemistry II	Unit I-Elimination and Free Radical Reactions Unit II-Oxidation and Reduction Reaction Unit III –Rearrangements Unit IV- Addition to Carbon Multiple Bonds Unit V-Reagents and Modern Synthetic Reactions	Global	50%change
П	Physical Chemistry I	Unit I- Quantum Mechanics I-1.3,1.6-1.9 deleted Unit II- Quantum Mechanics II Unit III- Quantum Mechanics III-combined four units and modified to three units Unit IV- Molecular Spectroscopy I-Shifted Raman Spectroscopy Unit V- Photochemistry	Global	30% change
II	Materiel Science	Unit I Optoelectronic materials Unit II Ceramic Materials Unit III Polymeric Materials Unit IV Composite Materials Unit V New Materials	National	100%chang
II	Bio inorganic Chemistry	Unit I- Essential trace elements, Oxygen transport and uptake proteins. Unit II- Electron transport proteins Unit III- Metal containing enzyme Unit IV- Biological redox enzyme Unit V- Metals and chelation in medicine	Global	70% change

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	Inorganic	Unit I- Electronic Spectroscopy		1
III	Chemistry II	Unit II- Optical rotatory dispersion, circular	Global	10%change
		dichroism and mossbauer spectoscopy		
	Fineson - Ju	Unit III-Nuclear magnetic resonance and	(3)	
		Electron paramagnetic resonance spectroscopy		TOWNS OF
		Unit IV-Organometallic Chemistry I		
		Unit V-Organometallic Chemistry II		
		5.12 Olefin isomerization using Ni catalyst -		
		Olefin metathesis catalysed by Schrock type		
		carbene deleted from the old syllabus.	1 194	
III	Physical	Unit I- Group Theory I	Global	25 %
	Chemistry II	Unit II- Group Theory I		change
		Unit III- Chemical Kinetics I - 4.5,4.6 module		
		deleted & combined two units to one unit.		
		Unit IV- Molecular Spectroscopy II		
		Unit V- Molecular Spectroscopy III- Shifted		
		this unit from semester II	1	FW M.
III	Chemistry of	Unit I- Alkaloids	Global	50%
	Natural product	Unit-II Terpenoids and Carotenoids	I THE T	change
		Unit III- Anthocyanines and Flavones		
		Unit IV- Purines and Steroids		
		Unit V- Aminoacids and Proteins		
III	Green and Supra	Unit-1 Green chemistry –I	National	20%
	Molecular	Unit -2 Green chemistry –II		change
	Chemistry	Unit-3 Supramolecular chemistry		
		Unit-4 Green chemistry-III		
	. 7	Unit-5 Graphene Functionalisation		
IV	Organic	Unit I-UV-Visible and IR Spectroscopy	Global	40%
	chemistry III	Unit II- NMR Spectroscopy		change
		Unit III- Planning an organic synthesis and	130 2	
		control elements		

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		Unit V- Pericyclic Reactions	T	Table
		Unit V - Organic Photo chemistry		
īV	Inorganic	Unit I-Coordination Chemistry		
	Chemistry III	Unit II-Stability of complexes and design of	Global	80%
		specialized ligands		change
		Unit III-Non-aqueous solvents		
		Unit IV-Chemistry of d-block and f-block		
		elements.		Total Black
	Man -Le	Unit V-Electron transfer and photochemical		
		reactions in complex		
IV	Physical	Unit I- Surface Chemistry -combined	Global	40% change
	chemistry III	unit II&III of old syllabus and modified		
		Unit II-Kinetics of fast reactions and catalysis-		
		2.7 to 2.11 from unit II of old syllabus.		
		Unit III-Irreversible Thermodynamics		
		Unit IV- Statistical Thermodynamics		
		Unit V-Polymer Chemistry -deleted 5.3 to 5.6		
		and added Dendrimers 5.6 to 5.10		
IV	Research	I. Literature Searching and	Global	60% change
	Methedology	Preparation of Project Report		
		II. Statistical Analysis and		
		Cheminformatics		
		III. Artificial Intelligence and its		
		application in Chemistry		
		IV. Experiments on Analytical		
		chemistry		
		v. Intellectual Property Rights and	10.40	
		Plagiarism		

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Practicals

m	Inorganic Practical-II	Gravimetric Estimations and Preperation of Inorganic metal Complexes	Global	No change
III	Organic Practical –II	Estimation of organic substance and double stage preparation of a solid compound	Global	30% change
IV	Physical Practical	Physical Chemistry Experiments	Global	20% change

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